

A Stability Indicating UV-HPLC Method for the Determination of Potential Impurities in Nandrolone Phenylpropionate Active pharmaceutical ingredient

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ABSTRACT

A reverse phase stability indicating HPLC method has been developed for the identification and quantification of potential impurities in Nandrolone phenylpropionate active pharmaceutical ingredient. The chromatographic separation of potential impurities was achieved in Inertsil ODS-3V column (250 mm x 4.6 mm, 5 µm) using gradient elution method. Mobile phase-A was prepared using 0.1% orthophosphoric acid in water and methanol mixture in a ratio of (90:10, v/v) and mobile phase-B was only acetonitrile. The developed method was validated as per the ICH guidelines for specificity, linearity, precision and accuracy. Specificity of the method was confirmed by peak purity analysis using photodiode array (PDA) detector. The value of correlation coefficient was greater than 0.999 for Nandrolone phenylpropionate and its six impurities. Accuracy of the method was established between 93.3% to 109.0% for all impurities. Nandrolone was found to be the major degradation impurity. The proposed method is suitable for routine as well as stability studies.

Keywords: Nandrolone phenylpropionate, nandrolone, anabolic, steroid.

INTRODUCTION

Nandrolone phenylpropionate (NPP) is known as 19-nortestosterone β-phenyl propionate¹. It is an anabolic-androgenic steroid. It is a fast-acting ester derivative form of Nandrolone and has strong anabolic effect with weak androgenic

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effect. NPP is injected intramuscularly every once in a week for 12 weeks and the dose may vary from 25 mg/mL to 50 mg/mL of steroid dissolves in oily formulation². Nandrolone Esters have been used in the treatment of osteoporosis, anemia, increase in muscle mass, to induce protein synthesis in skeletal muscles and improve respiratory muscle functions in COPD (Chronic Obstructive Pulmonary Disease)³⁻⁷. The effect of NPP in rats has been reported by Hale (1972)⁸.

The related substances method by TLC and assay method by UV spectrophotometer have been reported in Indian Pharmacopoeia and United States Pharmacopoeia respectively^{9,10}. The work on the development and validation of multi-residue method for the detection of wide range of hormonal anabolic compounds in hair using gas-chromatography-tandem mass spectrometry has been carried out¹¹. The effect of Nandrolone esters in the growth and histology of urine and hair has been studied by Groot et al¹². The validated GC-MS and UHPLC-MS method have been reported for the quantitative determination of anabolic steroids in formulation products and dietary/nutritional supplements¹³⁻¹⁶. Findings on the effect of side chain, the injection site and the injection volume on the pharmacokinetics and pharmacodynamics of Nandrolone esters in an oil vehicle in men is also available¹⁷. Various research articles have also reported for the development and analysis of Nandrolone esters¹⁸⁻²¹.

However, reverse phase HPLC method has not been reported yet for the determination of potential impurities in NPP (including major pharmacopoeias such as USP, Ph.Eur, BP, JP, and IP). Therefore, it is important to have a stability indicating method for the quantification of potential impurities in NPP which is specific and accurate. The quality, efficacy and safety are significantly affected by the impurities present in the drug product²². As per the current ICH Guideline of stability testing of drug substances, forced degradation must be carried out to establish that the method is stability indicating and the analytical methods need to be validated²³⁻²⁶. The current work involves method development, forced degradation and method validation for NPP.

METHODOLOGY

Materials and chemicals

NPP sample (99.7% purity, 10 g, Batch number: NPP/009/005), Imp-2 (98.3% purity, 1 g, Batch number: ND/031/016), Imp-3 (98.7% purity, 500 mg, Batch number: BD/035/009), Imp-4 (98.5% purity, 500 mg, Batch number: NPP/009/018), Imp-5 (98.9% purity, 500 mg, Batch number: NPP/009/023), Imp-6 (98.1% purity, 500 mg, Batch number: NPP/009/029) were obtained from chemical research division, Ipca laboratories Ltd. (Vadodara, India). Imp-1

(99.0% purity, 100 g, Lot number: STBD4764V) was obtained from Sigma-Aldrich (Steinheim, Germany). HPLC grade water for the analysis purpose was obtained from Milli-Q plus water purification system (Millipore, Bedford, MA, USA). HPLC grade acetonitrile and methanol was purchased from Merck India (Mumbai, India).

Instrumentation and HPLC conditions

The instruments used in the study include: HPLC separation module (2695) with UV detector (2487) and photodiode array detector (2996) (Waters Corporation, Milford, MA, USA); A photostability chamber model NEC-104RTS (Newtronic, Mumbai, India).

Inertsil ODS-3V column (250 mm x 4.6 mm, 5 μ m) HPLC column was used in the analysis (GLSciences). For the gradient elution method, mobile phase-A and mobile phase-B were used. Mobile phase-A was prepared using 0.1% orthophosphoric acid in water and methanol mixture in a ratio of (90:10, v/v) and mobile phase-B was only acetonitrile. The column temperature was kept at 40°C for the analysis; flow rate of 1.0 mL/min; injection volume was 20 μ L; detector wavelength was fixed at 240 nm and 210 nm; sample temperature was 25°C. Diluent was a mixture of water and acetonitrile in a ratio of (20:80, v/v). Gradient program is tabulated in Table 1.

Table 1. Linear gradient program.

Time/minutes	Mobile phase-A, %	Mobile phase-B, %
0	45	55
25	20	50
35	10	90
40	45	55
50	45	55

Preparation of sample and stock solution for validation

NPP test sample was prepared by transferring about 20 mg of sample into 50 mL volumetric flask, added 10 mL of acetonitrile to it and sonicated for 1.0 min and made up to the volume with diluent (400 μ g/mL).

Individual stock solution of about 800 μ g/mL was prepared by transferring about 20 mg each of NPP, imp-1, imp-2, imp-3, imp-4, imp-5 and imp-6 into separate 25 mL volumetric flasks, added 10 mL of acetonitrile to it and sonicated for 1.0 min and made up to the volume with diluent. From the above individual stock solution transfer 0.75 mL into separate 20 mL volumetric flasks, made up to the volume with diluent and shake well. This solution was labeled as standard stock solution (30.0 μ g/mL). Further from this standard stock solution, the desired solutions of different concentrations were prepared for validation.

Preparation of forced degradation samples

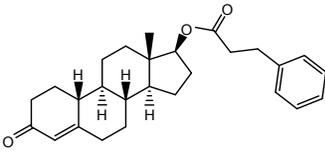
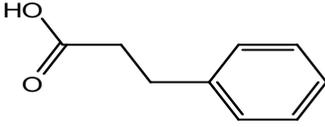
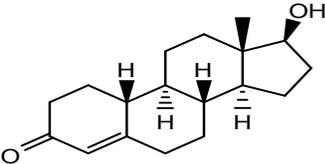
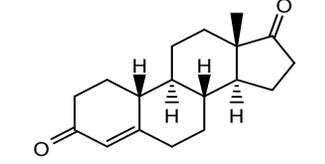
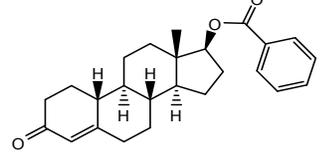
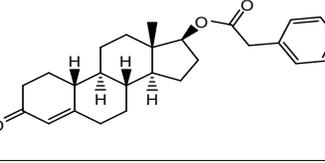
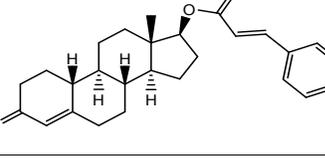
About 20 mg each of sample was weighed and transferred into 3 separate 50 mL volumetric flasks and labeled as 1, 2 and 3. 10 mL of acetonitrile was added into each volumetric and the sample was dissolved by sonicated for 1.0 min. 2 mL of 0.5 N hydrochloric acid solution, 5.0 % hydrogen peroxide solution and 0.5 N sodium hydroxide was also added into each volumetric flask. The flasks were kept in a water bath at 60°C for 2 h (acid hydrolysis), 60°C for 6 h (oxidative degradation) and at room temperature for 1 h (base hydrolysis) respectively. The excess of acid or base in volumetric flask 1 and 3 were neutralized and made upto the mark with diluents. Corresponding blank solutions were prepared. Thermal degradation was performed on solid NPP sample at 70°C for 48h. Photolytic degradation was performed by spreading the sample on petri dish and kept in a photostability chamber model NEC-104RTS (Newtronic) to get the light intensity of 1.2 million Lux hours for white light and 200 Wh/m² for ultraviolet region.

RESULTS and DISCUSSION

HPLC method development

The literature search was done to check the method availability but little information was available regarding the quantification of related impurities of NPP. Hence it was decided to initiate the reverse phase HPLC method development. The aim of the proposed method is to achieve the baseline separation between all related impurities and NPP. Structure and details of NPP and its impurities are listed in Table 2. The λ_{\max} of NPP and its potential impurities were showed in Figure 1.

Table 2. Structure and details of NPP and its impurities.

S.No	Name	Structure	Code	Source
1	Nandrolone phenylpropionate		NPP	Drug
2	3-Phenylpropionic acid		Imp-1	Degradation Impurity
3	Nandrolone		Imp-2	Raw material and degradation Impurity
4	Bolandione		Imp-3	Process impurity
5	Nandrolone benzoate		Imp-4	Process impurity
6	Nandrolone phenyl acetate		Imp-5	Process impurity
7	Nandrolone phenylcinnamoate		Imp-6	Process impurity

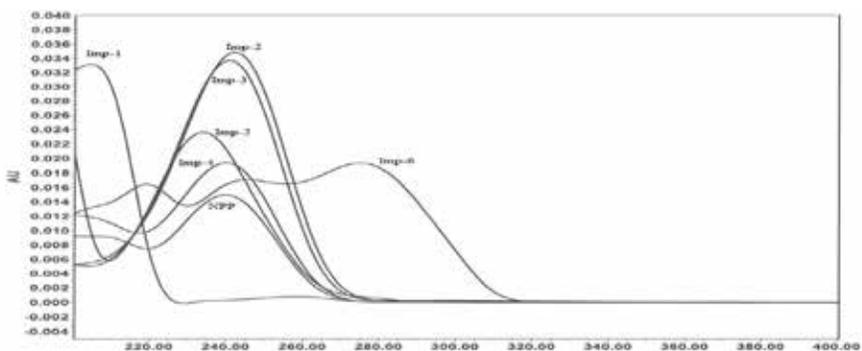


Figure 1. The λ_{\max} of NPP and its potential impurities.

The method development was initiated using water and acetonitrile as mobile phase-A and Mobile phase-B in gradient elution method, using C-8 column with dimension (150 mm x 4.6 mm, 5 μ m). The peak of NPP and imp-6 co-elutes and the peak shape of imp-1 was found distorted. Finally after conducting several experiments the method was finally developed using linear gradient program. For the gradient elution method, mobile phase-A and mobile phase-B were used. Mobile phase-A was prepared using 0.1% orthophosphoric acid in water and methanol mixture in a ratio of (90:10, v/v) and mobile phase-B was only acetonitrile. Inertsil ODS-3V column (250 mm x 4.6 mm, 5 μ m) was used. The column temperature was kept at 40°C and detector wavelength was selected as 240 nm and 210 nm (210 nm for imp-1) throughout the analysis. The NPP peak eluted at about 26 min with the base line separation of all impurities (Figure 2 and 3).

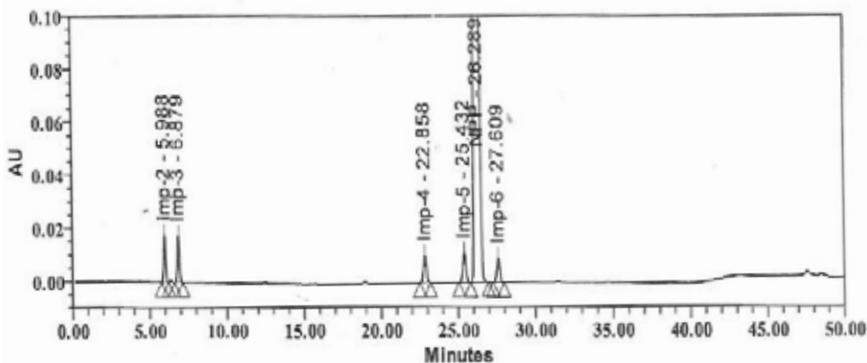


Figure 2. Chromatogram of spiked impurities in NPP at wavelength 254 nm.

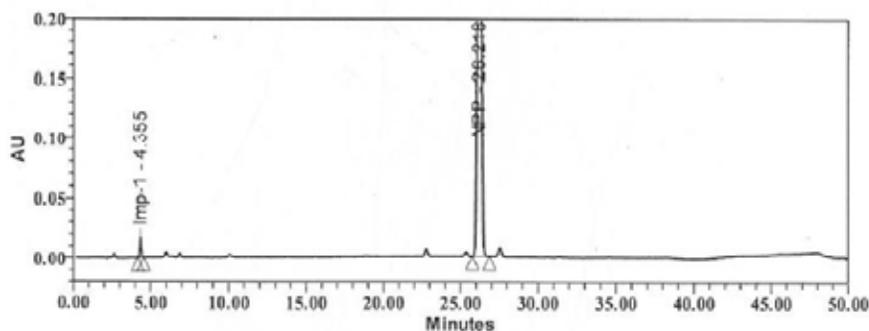


Figure 3. Chromatogram of spiked impurities in NPP at wavelength 210 nm.

Forced degradation study

The forced degradation of NPP sample was conducted in acidic, basic, thermal, oxidative and photolytic conditions. The significant degradation of NPP was observed in acidic and basic conditions. In both the conditions, the impurity formed was identified as imp-2 at 240 nm and imp-1 in 210 nm. There was no impact of oxidative, thermal and photolytic conditions on NPP. The peak purity analysis of the analyte peak obtained from PDA detector in all stress samples confirmed the spectral purity of NPP peak demonstrates the stability indicating capability of the method. The chromatograms are shown in Figure 4 and 5 and data is tabulated in Table 3.

Table 3. Percentage of degradation under different conditions using NPP sample at 400 µg/mL.

S No.	Condition	Degradation %
1	Acidic (0.5N HCl, 60°C, 2 h)	3.5
2	Basic (0.5N NaOH, room temperature, 1 h)	6.7
3	Oxidation (5% H ₂ O ₂ , 60°C, 6 h)	-
4	Thermal (70°C, 48 h)	-
5	Photolytic degradation	-

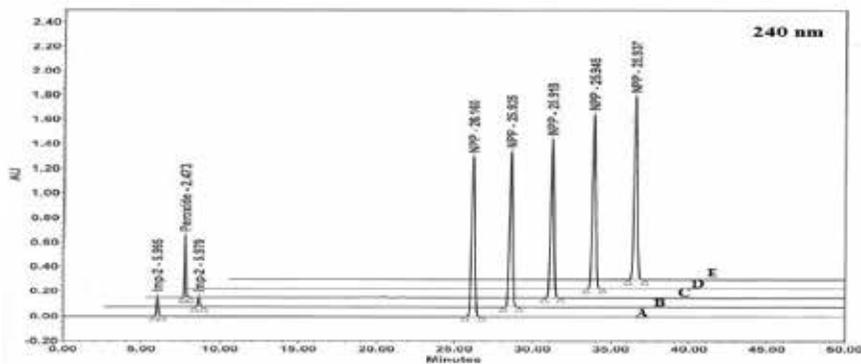


Figure 4. Chromatogram of forced degradation studies of NPP at 240 nm: (A) base hydrolysis, (B) acid degradation (C) oxidative degradation, (D) thermal degradation, (E) photolytic degradation.

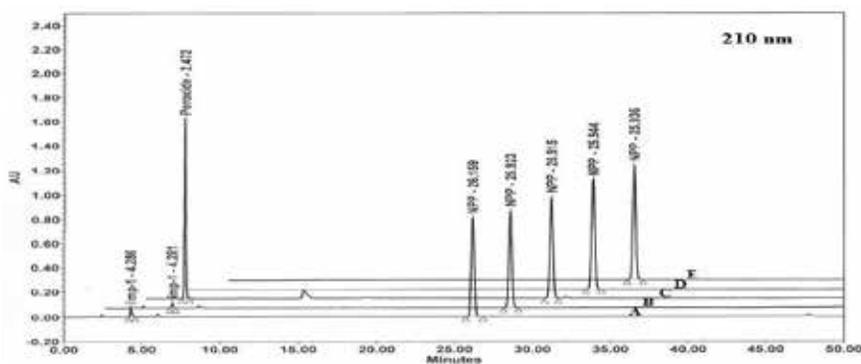


Figure 5. Chromatogram of forced degradation studies of NPP at 210 nm: (A) base hydrolysis, (B) acid degradation (C) oxidative degradation, (D) thermal degradation, (E) photolytic degradation.

Method validation

As per ICH guidelines the method must be validated to establish specificity, solution stability, sensitivity, linearity, precision, accuracy and robustness. The method validation was carried out for imp-1, imp-2, imp-3, imp-4, imp-5, imp-6 and NPP. A summarized result of method validation is tabulated in Table 4.

Table 4. Method Validation summary data

Parameter	Imp-1	Imp-2	Imp-3	Imp-4	Imp-5	Imp-6	NPP
Retention Time (RT)	4.36	5.99	6.88	22.86	25.43	27.61	26.29
Relative RT	0.17	0.23	0.26	0.87	0.97	1.05	1.00
Resolution	76.00	3.85	53.85	6.97	2.27	3.50	-
Symmetry factor	1.04	1.03	1.02	1.00	1.00	1.01	1.03
Response factor	0.56	0.71	0.69	0.77	0.68	0.88	1.00
Linearity	0.9998	0.9991	0.9997	0.9997	0.9993	0.9997	0.9996
Detection Limit ($\mu\text{g/mL}$)	0.0721	0.0362	0.0364	0.0360	0.0358	0.0359	0.0361
Quantitation Limit ($\mu\text{g/mL}$)	0.1503	0.1205	0.1212	0.1199	0.1193	0.1195	0.1202
Intra-day precision (n=6, % RSD)	1.95	0.47	0.57	1.14	0.60	0.83	--
Inter-day precision (n=6, % RSD)	0.92	0.66	1.32	1.57	1.48	1.70	-
Accuracy at LOQ (n=3, %)	94.6	106.7	103.3	104.4	108.0	93.3	--
Accuracy at 100 (n=3, %)	102.7	108.1	107.8	105.9	105.9	105.9	--
Accuracy at 150 (n=3, %)	103.9	106.3	109.0	106.0	105.4	104.5	--

Specificity

Specificity is the ability of chromatographic method to separate the analyte peak from blank interference, impurities and degradants. Specificity was done by injecting blank, NPP sample and NPP sample Co-spiked with impurities. The NPP peak was found well resolved from all six impurities and no blank interference was observed. Forced degradation study was also performed to demonstrate method specificity as well as stability indicating capability of the proposed method.

Solution stability

Stability of solutions was established by injecting spiked and unspiked sample solution kept in HPLC vial at 25°C in auto sampler. Area of each impurity was checked after 24 hrs against initial (0 h) and the % difference was found less than 10%, hence sample solution are stable for 24 h at 25°C.

Limit of detection and quantitation (Sensitivity)

The detection Limit (LOD) and Quantitation limit (LOQ) determines the sensitivity of the proposed method. The LOD for each impurity (imp-1, imp-2, imp-3, imp-4, imp-5 and imp-6) and NPP were calculated and found to be in the range of 0.0358 $\mu\text{g/mL}$ to 0.0721 $\mu\text{g/mL}$. The LOQ for each impurity (imp-1, imp-2, imp-3, imp-4, imp-5 and imp-6) and NPP were calculated and found to be in the range of 0.1193 $\mu\text{g/mL}$ to 0.1503 $\mu\text{g/mL}$.

Linearity and range

Linearity determines the ability of the analytical method to get the test results that are directly proportional to the analyte concentration within the given range. The solutions were prepared at six different concentration levels for all specified impurities and sample from LOQ level to 150% of impurity limit [0.15% of the drug concentration (400 µg/mL)]. The result of correlation coefficient obtained for all analyte were found to be greater than 0.999, confirms positive correlation between peak area and concentration of impurities and NPP peak (Figure 6).

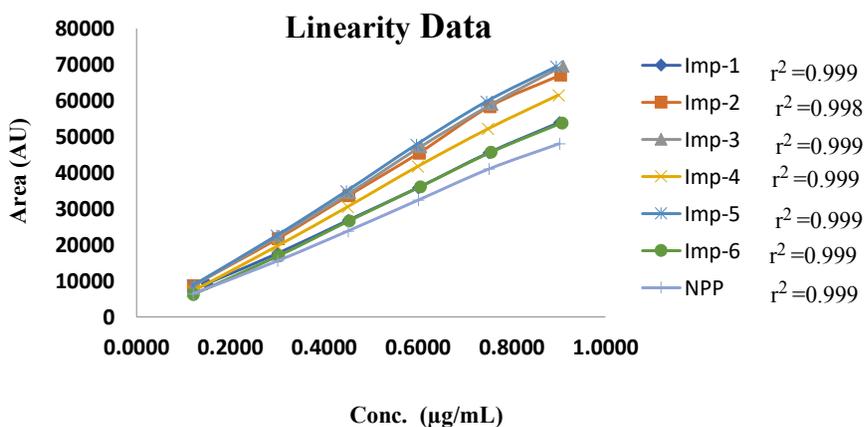


Figure 6. Linearity data of all 6-impurities and NPP.

Accuracy

The accuracy of the method was expressed in the term of percentage recovery. The accuracy of the present method was performed at LOQ level, 100% level (0.15% of the drug substances) and 150% level (0.225% of the drug substances). The experiment was performed in triplicate. The percentage recovery of all impurities ranged between 93.3% to 109.0% indicating the accuracy of the method.

Precision

The precision (Intra-day precision) of the method was evaluated by preparing six individual preparations of NPP sample and spiked with 0.15 % (0.6 µg/mL) of each impurity with respect to sample concentration (400 µg/mL). Peak area of each impurity was checked and percentage relative standard deviation was calculated and found less than 2.0 % for each impurity.

The precision (inter-day precision) was also determined by performing the same procedure in same lab by different analyst on different instrument and on different dates. The percentage RSD for each impurity was found below 2.0% in both the precision indicated the good precision of the proposed method.

Robustness

The robustness of the analytical method was checked by evaluating the influence of small deliberate modifications in the HPLC method parameter. The studies showed that there was no impact on the method even by deliberately changing the chromatographic parameters (i.e. column temperature, flow rate and mobile phase composition). It was observed that the resolution among all the impurities and NPP peak was higher than 2.0. This showed the reliability of the method during routine usage of the method.

The proposed method was found to be better and advantageous over previous reported methods for quantification of potential impurities as compared to the TLC and UV spectrophotometer method describes in Indian pharmacopoeia and United States Pharmacopoeia respectively. The proposed method has been validated as per the current ICH guidelines and easy to use for routine analysis as well as stability studies.

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