

SYNTHESIS OF SOME 4-PYRROLYLPHENYLTHIAZOLE DERIVATIVES AND INVESTIGATION
OF THEIR ANTIMICROBIAL ACTIVITIES

BAZI 4-PIROLİL FENİL TİYAZOL TÜREVLERİNİN SENTEZİ VE ANTİMİKROBİYAL
ETKİLERİNİN ARAŞTIRILMASI

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In this study, some 2-substituted 4-[4-(pyrrol-1-yl)phenyl]thiazole derivatives were obtained reacting some 1,4-dicarbonyl compounds and 2-substituted 4-(4-aminophenyl)thiazoles. The latter were prepared by hydrolyzing of 2-substituted 4-(4-acetylaminophenyl)thiazoles. The antibacterial and antifungal activities of the pyrrole compounds were investigated and no considerable activity was obtained.

Bu çalışmada, 4-(α -kloroasetil)asetanilid ile tiyoasetamid veya etiyonamid'in reaksiyona sokulmasıyla elde edilen 2-süstitüe 4-(4-asetilaminofenil)tiyazol türevleri hidroliz edilerek 2-süstitüe 4-(4-aminofenil)tiyazol türevleri sentezlenmiştir. Bu amino türevleriyle bazı 1,4-dikarbonil bileşiklerinin asetik asid içerisinde ısıtılmasıyla 2-süstitüe 4-(4-pirolilfenil)tiyazol türevlerine ulaşılmıştır. Pirol türevi bileşiklerin antibakteriyel ve antifungal aktiviteleri araştırılmış ve kaydedeğer bir ekti elde edilememiştir.

Keywords : Pyrrole; Thiazole; Antibacterial; Antifungal effects

Anahtar kelimeler: Pirol; Tiyazol; Antibakteriyel ; antifungal aktivite

Introduction

The chemistry of pyrrole has been the subject of investigations by an ever increasing number of researchers since the discovery that the pyrrole ring was a part of haemin and of chlorophyll molecules. A large number of pyrrole derivatives having antibiotic properties have been isolated from microbial sources (1,2).

Thiazole nucleus is also found in a variety of naturally occurring products which exert varied pharmacological effects(3). These observations urged us to synthesize some 4-[4-(pyrrol-1-yl)phenyl]thiazole derivatives and to test their antimicrobial activities.

Materials and Methods

Melting points were determined by using a Gallenkamp apparatus and are uncorrected. Spectroscopic data were recorded on the following instruments: IR: Shimadzu IR 435 spectrophotometer; ¹H-NMR: Bruker DPX 400 and Jeol 60 NMR spectrometers; MS: VG Platform mass spectrometer. Analyses for C, H, N were within 0.4% of the theoretical values. 4-(α -Chloroacetyl)-acetanilide(4), 1-phenyl-3-carbethoxy-1,4-pentadione(5), compounds **3a** and **4a** (6) were prepared according to the methods reported in the literature. Some characteristics of the compounds are reported in Table 1.

2-Substituted 4-(4-acetylaminophenyl)thiazoles, **3a,b** General Procedure

A mixture of **2** (5 mmol) and an appropriate thioamide, **1a** or **1b**, in ethanol (100 ml) was refluxed for 2h. The cooled solution was neutralized with NaHCO₃. The precipitate formed was filtered, washed with water and crystallized from ethanol.

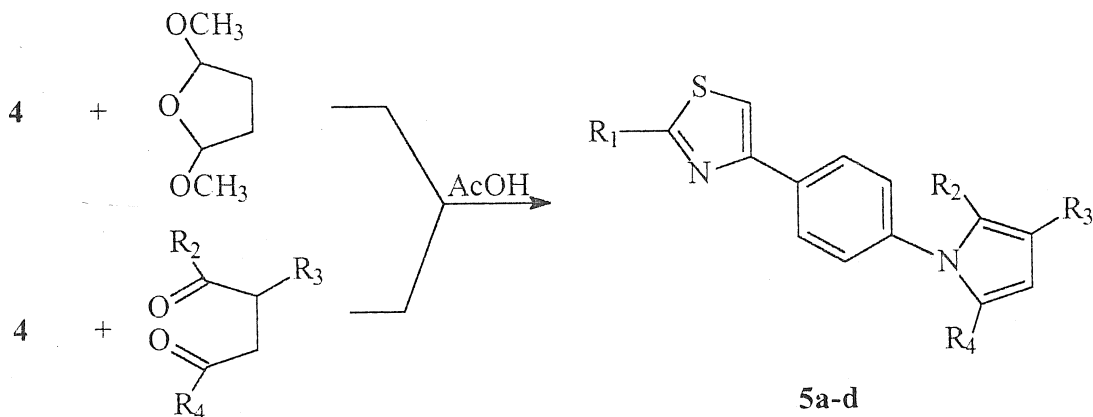
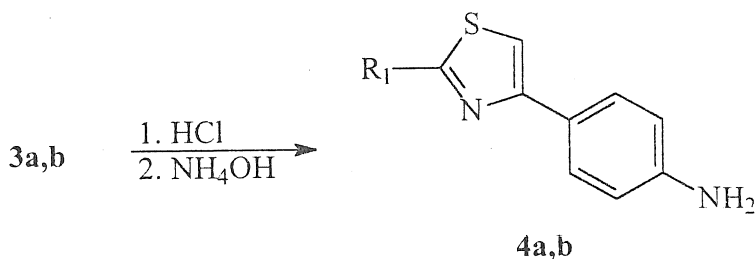
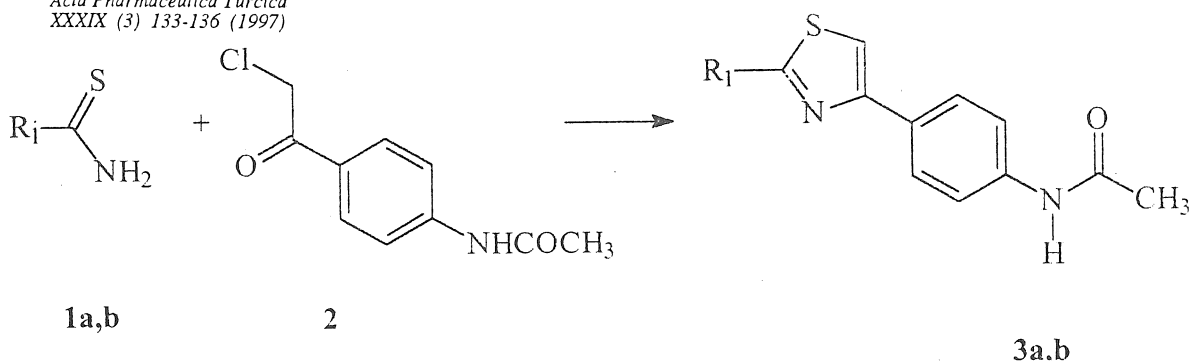
3a : IR (KBr, cm⁻¹) 3399 (N-H), 1655 (C=O), 1613-1471 (C=N, C=C); ¹H-NMR (60MHz, DMSO-d₆, σ ppm) 2.05 (3H, s, COCH₃), 2.65 (3H, s, thiazole-2-CH₃), 7.52-8.05 (5H, m, Ar-H and thiazole-5-H), 10 (1H, s, N-H).

3b : IR (KBr, cm⁻¹) 3292 (N-H), 1666 (C=O), 1596-1468 (C=N, C=C); ¹H-NMR (90 MHz, DMSO-d₆, σ ppm) 1.30 (3H, t, CH₂-CH₃), 2.05 (3H, s, COCH₃), 2.90 (2H, q, CH₂-CH₃), 5.50 (2H, brs, NH₂), 6.60 (2H, d, Ar-H), 7.50-7.80 (4H, m, Ar-H), 7.90 (1H, s, thiazole-5-H), 8.46 (1H, d, pyridyl-6-H), 10 (1H, brs, N-H).

2-Substituted 4-(4-aminophenyl)thiazoles, **4a,b** General Procedure

3a or **3b** (10 mmol) was dissolved in the mixture of ethanol (50 ml) and 4N HCl (50 ml). The solution obtained was refluxed for 1 h. The cooled solution was neutralized with NH₄OH solution. The precipitate was crystallized from diluted ethanol.

4a : IR (KBr, cm⁻¹) 3490, 3322, 3203 (N-H), 1634-1468 (C=N, C=C), 826 (1,4 disubstituted benzene); ¹H-NMR 60MHz, DMSO-d₆, σ ppm) 2.63 (3H, s,



thiazole-2-CH₃), 4.95 (2H, s, NH₂), 6.55 (2H, d, Ar-H), 7.32 (1H, s, thiazole-5-H), 7.58 (2H, d, Ar-H).

4b: IR (KBr, cm⁻¹) 3428, 3320 (N-H), 1644-1460 (C=N, C=C), 826 (1,4-disubstituted benzene); ¹H-NMR (400 MHz, DMSO-d₆, σppm) 1.28 (3H, t, CH₂-CH₃), 2.85 (2H, q, CH₂-CH₃), 5.34 (2H, brs, NH₂), 6.66 (2H, d, J: 8.32 Hz, Ar-H), 7.65-7.79 (4H, m, Ar-H, pyridyl-3-H and 5-H), 7.89 (1H, s, thiazole-5-H), 8.60 (1H, d, j:5.1 Hz, pyridyl-6-H).

*-Substituted 4-[4-(pyrrol-1-yl)phenyl]thiazoles,
5a-d General Procedure*

A mixture of **4a** or **4b** (5 mmol) and an appropriate dicarbonyl compound (5.2 mmol) in acetic acid (25 ml) was stirred in water bath for 30 min. The solvent

2 was evaporated under vacuum. The residue was dissolved in water. The solution was neutralized with NaHCO₃. The precipitate was filtered and crystallized from ethanol.

5a: IR (KBr, cm⁻¹) 1622-1450 (C=N, C=C); ¹H-NMR (400 MHz, DMSO-d₆, σppm) 2.37 (3H, s, thiazole-2-CH₃), 6.29 (2H, s, pyrrole-3, 4-H), 7.42 (2H, s, pyrrole-2,5-H) 7.63 (2H, d, j:12Hz, Ar-H), 7.93 (1H, s, thiazole-5-H), 7.63 (2H, d, j:12Hz, Ar-H), 7.93 (1H, s, thiazole-5-H), 8.01 (2H, d, j:8.25 Hz, Ar-H); EI-MS: m/z :242 (M+2, 6%), 241.2 (M+1, 15%), 240.1 (M, 100%), 198.6, 170.9, 167, 153.9, 139.

5b: IR (KBr, cm⁻¹) 1600-1470 (C=N, C=C); ¹H-NMR (400 MHz, DMSO-d₆, σppm) 1.99 (6H, s, pyrrole-2,5-CH₃), 2.73 (3H, s, thiazole-2-CH₃), 5.81 (2H, s, pyrrole-3,4-H), 7.30 (2H, d, j:7.74 Hz, Ar-H), 8.00

Table 1. Some characteristics of the compounds

| Comp. | R ₁ | R ₂ | R ₃ | R ₄ | M.p. (°C) | Yield (%) | Mol. Formula (Mol. Weight) |
|-------|------------------|------------------|--|--------------------------------|--------------|--------------|---|
| 3b | | - | - | - | 194 | 58 | - |
| 4b | | - | - | - | 132 | 67 | - |
| 5a | -CH ₃ | -H | -H | -H | 162 | 48 | C ₁₄ H ₁₂ N ₂ S (240.3) |
| 5b | -CH ₃ | -CH ₃ | -H | -CH ₃ | 105 | 65 | C ₁₆ H ₁₆ N ₂ S (268.4) |
| 5c | | -H | -H | -H | 149 | 56 | C ₂₀ H ₁₇ N ₃ S (331.4) |
| 5d | | -CH ₃ | -CO ₂ C ₂ H ₅ | -C ₆ H ₅ | 145 | 55 | C ₃₀ H ₂₇ O ₂ S (451.6) |

Table 2. The MIC values of compounds 5a-d

| Comp. | S. aureus ATCC 25923 | E.coli ATCC 25922 | P.aeruginosa ATCC27.853 | C. albicans ATCC 10231 |
|--------------|-------------------------|----------------------|----------------------------|---------------------------|
| 5a | 125 | 62.5 | 62.5 | 125 |
| 5b | 125 | 62.5 | 62.5 | 62.5 |
| 5c | >125 | >125 | >125 | 62.5 |
| 5d | 125 | 62.5 | 62.5 | 62.5 |
| Ceftriaxone | 2 | 0.03 | 8 | 2 |
| Clotrimazole | - | - | - | 2 |

(1H, s, thiazole-5-H), 8.05 (2H, d, j:7.88 Hz, Ar-H).
5c : IR (KBr, cm⁻¹) 1620-1453 (C=N, C=C), 849
(1,4-disubstituted benzene); ¹H-NMR (400 MHz,
DMSO-d₆, σppm) 1.30 (3H, t, -CH₂-CH₃), 2.88 (2H,

q, -CH₂-CH₃), 6.31 (2H, s, pyrrole-3,4-H), 7.46 (2H,
s, pyrrole-2,5-H), 7.71 (2H, d, j:8.52 Hz, Ar-H), 7.79
(1H, d, j:5.03 Hz, pyridyl-5-H), 7.85 (1H, s,
pyridyl-3-H), 8.14 (2H, d, j: 8.52 Hz, Ar-H), 8.35 (1H,
s, thiazole-5-H), 8.64 (1H, d, j:5.15 Hz, pyridyl-6-

s, thiazole-5-H), 8.64 (1H, d, j :5.15 Hz, pyridyl-6-H).

5d: IR (KBr, cm^{-1}) 1681 (C=O), 1597-1444 (C=N, C=C), 1233, 1097 (C-O); $^1\text{H-NMR}$ (400 MHz, DMSO- d_6 , σ ppm) 1.23-1.32 (6H, m, two $\text{CH}_2\text{-CH}_3$), 2.37 (3H, s, pyrrol-2- CH_3), 2.87 (2H, q, pyridyl-2- $\text{CH}_2\text{-}$), 4.24 (2H, q, $-\text{OCH}_2\text{-}$), 6.71 (1H, s, pyrrol-4-H), 7.10-7.22 (5H, m, pyrrol-5- C_6H_5), 7.38 (2H, d, j :8.16 Hz, Ar-H), 7.78 (1H, d, j :5.0 Hz, pyridyl-5-H), 7.85 (1H, s, pyridyl-3-H), 8.15 (2H, d, j :8.21 Hz, Ar-H), 8.43 (1H, s, thiazole-5-H), 8.63 (1H, d, j :5.07 Hz, pyridyl-6-H); EI-MS m/z : 495.4 (M+2, 10%), 494.4 (M+1, 35%), 493.4 (M, 100%), 464.3, 448.3, 420, 368.2, 291.1, 286, 273, 265.1, 246.7, 232, 223.7.

Determination of the antimicrobial activity

Antibacterial and antifungal activities of the compounds **5a-d** were determined using the tube dilution technique (7-9). The stock solutions of the compounds were prepared in DMSO. Ceftriaxone and clotrimazole were used as control antibacterial and antifungal agents. The MIC values are given as $\mu\text{g/ml}$. The standard bacteria and fungi strains used and MIC values are shown in Table 2.

Results and Discussion

The first group of the compounds **3a,b** was obtained by reacting 4-(α -chloroacetyl) acetanilide and thioacetamide or ethionamide (2-ethyl-4-thiocarboxamidopyridine). The reaction is an application of the Hantzsch thiazole synthesis (1). In the second step, the acetamido group of the compounds **3a, b** was converted to amine by hydrolysing in hydrochloric acid solution. The last step involves Paal-Knorr pyrrole synthesis (3). The compounds **5a-d** were synthesized by heating the amine compounds **4a,b** and appropriate 1,4-dicarbonyl compounds.

The newly synthesized compounds were characterized by elemental analyses, IR, $^1\text{H-NMR}$ and MASS spectral data. Characteristic IR absorption band of the amide carbonyl moiety of the compounds **3a,b** at about 1660 cm^{-1} were absent in the spectra of the hydrolyzed products **4a,b** and pyrrole compounds **5a-d**. Similarly, the strong stretching bands at about $3400\text{-}3200\text{ cm}^{-1}$ originated from NH groups of the compounds **3** and **4** are no longer present in the IR spectra of the compounds **5a-d**. In the $^1\text{H-NMR}$ spectra, the signals

due to 5-H protons on the thiazole ring which are common in all compounds are shown as singlets at about 7.80 ppm. The protons of amide residue of the compounds **3a,b** are observed as broad singlets at about 10 ppm. The protons of amino group which formed after hydrolyzing of the amid group resonate as broad singlets as well at about 5 ppm. After cyclizing to pyrrole rings, the signals due to NH are no longer present in the compounds **5a-d**. Although 3-H protons of the pyridyl rings overlap with the other aromatic protons. 5,6-H Protons are observed as doublets. However, these doublets are easily distinguished with their characteristic j values (i.e. 5 Hz) from the other doublets.

Molecular ions corresponding to M^+ are observed in the EI-MS spectra of the compounds **5a** and **5d**. $\text{M}+1$ is also obtained from ES-MS spectra of the compounds **5d**.

Antibacterial and antifungal activities of the compounds under investigation were determined using tube dilution technique (7-9). Although no considerable antibacterial and antifungal effect could be observed, the most effective MIC value was found as $62.5\text{ }\mu\text{g/ml}$.

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